

Pergamon

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## A Suite of Odd and Even Carbon-Numbered Spiroacetals in Bactrocera latifrons. Synthesis and Stereochemistry.

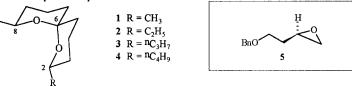
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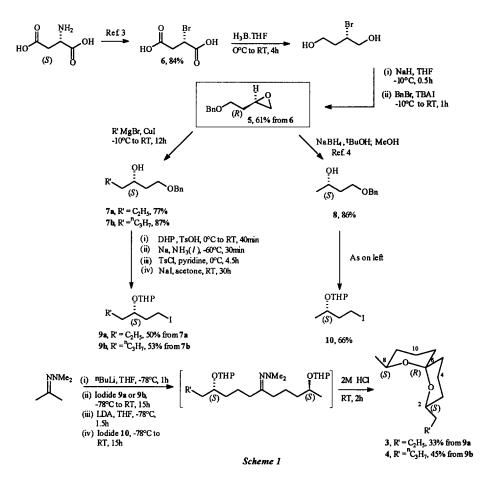
Abstract: Female abdominal tips from the pestiferous fruit-fly species, *Bactrocera latifrons* (Hendel) provide a suite of odd and even 2-alkyl-8-methyl-1,7-dioxaspiro[5.5] undecanes (alkyl = methyl, ethyl, "propyl, "butyl) which are shown by synthesis and enantioselective gas chromatography to possess the (2S, 6R, 8S) stereochemistry. © 1997 Elsevier Science Ltd.

Alkyl- and hydroxy-substituted spiroacetals have been identified from Tephritid fruit-fly species, particularly those from the genus *Bactrocera*.<sup>1</sup> With a few exceptions, alkyl spiroacetals contain nine, eleven or thirteen carbons in an unbranched arrangement, and any accompanying even carbon-numbered spiroacetal is of much lower relative abundance. Only in the case of 2-ethyl-8-methyl-1,7-dioxaspiro[5.5] undecane 2, from *B.nigrotibialus* (Perkins), has the sense of chirality been determined. The absolute stereochemistry of 2 was (2*S*, 6*R*, 8*S*) the same as for the major accompanying 2,8-dimethyl-1,7-dioxaspiro[5.5]undecane 1.<sup>2</sup> In view of our interest in insect-derived oxygen heterocycles and the necessity to furnish structural and stereochemical data for biosynthetic proposals, we now wish to report that females of the pest species *B.latifrons* (Hendel) provide a suite of four homologous 2-alkyl-8-methyl-1,7-dioxaspiro[5.5]undecanes, whose structures and absolute stereochemisty have been established by synthesis and enantioselective gas chromatography as (2S, 6R, 8S)-1, 2, 3 and 4 in accord with our previous prediction.<sup>2</sup>



A general synthetic approach to enantiomers of **3** - **4** utilises (*R*) - (2-benzyloxyethyl)oxirane **5**, now readily available from (*S*)-aspartic acid.<sup>3</sup> (Both (*S*) and (*R*)-aspartic acids are commercially available). This procedure, in which **5** is the convenient precursor of both required alkylating agents 7 and **8**, would also furnish enantiomers **1** and **2**, but we had already synthesised these by other means.<sup>2</sup> The steps are outlined in Scheme 1, and we found that reduction of the bromo-succinic acid **6** was better behaved with freshly prepared H<sub>3</sub>B.THF. Reductive opening of the oxirane **5** with NaBH<sub>4</sub> in 'BuOH-MeOH proceeds without removal of the benzyl group.<sup>4</sup> Spiroacetals **3** and **4**<sup>5</sup> acquired in this way possessed (2*S*, 6*R*, 8*S*) stereochemistry with >99.5%ee, based on enantioselective gas chromatography. Optical rotations for **3** and **4** were consistent with those previously reported for **1** and **2**.<sup>6</sup>

Racemic 3 and 4 resulted from use of racemic (2-benzyloxethyl)oxirane, formed by epoxidation of but-3en-1-ol. (Racemic 1 and 2 were already available).<sup>2</sup> Examination of a hexane abdominal tip extract of female *B.latifrons* showed the four spiroacetals 1 - 4 were the (*E,E*) diastereomers with (2*S*,6*R*,8*S*) stereochemistry, and ee >95% for 1, 2, and 4 and ee 91% for 3. It is of interest to note that one alkyl group is unchanging, with variation in the other from methyl to <sup>n</sup>butyl.



Although the spiroacetals 1 - 4 are present in comparable amounts, overall they are minor components of the abdominal tip extracts. A detailed analysis of female and male rectal components from *B. latifrons* has recently been reported.<sup>7</sup> Isomers 2 and 4 were previously reported<sup>8</sup> as very low level components from the rectal gland of **male** *B.latifrons* and it is very likely the (2S, 6R, 8S) stereochemistry applies also. The biological role of these components is currently being assessed.

## REFERENCES

- 1. Fletcher, M.T.; Kitching, W. Chem. Revs., 1995, 95, 789.
- 2. Perkins, M.V.; Kitching, W.; König, W.A.; Drew, R.A.I. J. Chem. Soc. Perkin Trans. 1, 1990, 2501.
- 3. Frick, J.A.; Klassen, J.B.; Bathe, A.; Abramson, J.M.; Rappoport, H. Synthesis, 1992, 621.
- 4. Okawa, A.; Hiratsuka, H.; Soai, K. Bull. Chem. Soc. Japan, 1987, 60, 1813.
- 5. All new compounds exhibited satisfactory HRMS for M<sup>4</sup> and concordant <sup>1</sup>H and <sup>13</sup>C nmr spectra. <sup>13</sup>C nmr (CDCl<sub>3</sub>, 125 MHz) 3: 96.0, 68.7, 65.1, 38.7, 35.5, 35.3, 32.9, 31.4, 21.8, 19.1, 19.0, 18.9, 14.2. 4: 96.0, 68.9, 65.0, 36.2, 35.5, 35.3, 32.9, 31.4, 28.1, 22.9, 21.8, 19.0, 18.9, 14.1. Optical rotations: 3,  $[\alpha]_D^{23} 40.5^\circ$  (c, 1.08, pentane) and 4,  $[\alpha]_D^{23} 53.0^\circ$  (c, 0.67, pentane).
- and 4,  $[\alpha]_D^{23} = 53.0^\circ$  (c, 0.67, pentane). 6. Optical rotations: 1,  $[\alpha]_D^{23} = 58.7^\circ$  (c, 1.6 in pentane) and 2,  $[\alpha]_D^{23} = 72.9^\circ$  (c, 0.38 in pentane). See also (a) Mori, K.; Watanabe, H. *Tetrahedron*, 1986, 42, 295 and ref. 2 and (b) Mori, K.; Ikunaka, M. *Liebigs Ann. Chem.*, 1987, 333.
- 7. Avery, J.W.; Liquido, N.; Cunningham, R.T.; Leonhardt, B.A.; Waters, R.M. J. Ent. Sci., 1996, submitted.
- Kitching, W.; Lewis, J.A.; Fletcher, M.T.; Drew, R.A.I.; Moore, C.J.; Francke, W. J. Chem Soc. Chem. Commun., 1986, 853.

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